



(University of Choice)

**MASINDE MULIRO UNIVERSITY OF  
SCIENCE AND TECHNOLOGY  
(MMUST)**

**UNIVERSITY EXAMINATIONS**

**2021/2022 ACADEMIC YEAR**

**SECOND YEAR SECOND SEMESTER  
MAIN EXAMINATIONS**

**FOR THE DEGREE OF  
B.SC RENEWABLE ENERGY AND BIOFUELS TECHNOLOGY**

**COURSE CODE: RET 272**

**COURSE TITLE: THERMODYNAMICS I**

**DURATION: 2 HOURS**

**DATE: 20-4-2022**

**TIME: 12.00-14.00 HRS**

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**INSTRUCTIONS TO CANDIDATES**

- (i) Answer **Question 1 (Compulsory)** and any other **TWO** questions
- (ii) All symbols have their usual meaning
- (iii) Use steam tables provided

This paper consists of **3** printed pages. Please Turn Over



MMUST observes **ZERO** tolerance to examination cheating

**QUESTION ONE (Compulsory) – 30 Marks**

- a) State the Non-Flow Energy Equation. (4 Marks)
- b) Steam at 110 bar has a specific volume of 0.0196 m<sup>3</sup>/kg. Find the following properties:
- i) Temperature. (3 Marks)
  - ii) Internal energy. (3 Marks)
- c) Given steam at 0.5MPa with an enthalpy of 2.4MJ/kg, determine the:
- (i) Dryness fraction. (5 Marks)
  - (ii) Specific volume. (2 Marks)
  - (iii) Internal energy. (2 Marks)
- d) Show that for a perfect gas the following specific heats can be expressed as shown below:
- (i)  $C_v = \frac{R}{\gamma - 1}$  (4 Marks)
  - (ii)  $C_p = \frac{\gamma R}{\gamma - 1}$  (3 Marks)
- (e) Give two conditions for a thermodynamic equilibrium. (4 Marks)

**QUESTION TWO (20 Marks)**

- a) Show that for a working fluid undergoing an adiabatic process, the work done can be expressed as:

$$W = \frac{P_1 V_1 - P_2 V_2}{\gamma - 1}$$

(7 Marks)

- b) Air at 1bar and 20<sup>0</sup>C initially occupying a cylinder volume of 0.016m<sup>3</sup> is compressed reversibly and adiabatically by a piston to a pressure of 7 bar.

Calculate the:

- i) Mass of air. (2 Marks)
- ii) Final temperature. (3 Marks)

iii) Final specific volume.

**(6 Marks)**

iv) Net work done.

**(2 Marks)**

**QUESTION THREE (20 Marks)**

Steam at 70 bar, 300°C is contained in a rigid cylinder of volume 0.02m<sup>3</sup>. The cylinder is cooled until the pressure is 40 bar.

a) Determine the state of steam after cooling

**(6 Marks)**

b) Calculate the amount of heat rejected by steam

**(9 Marks)**

c) Show the process on a t-s diagram indicating the area which represents heat flow

**(5 Marks)**

**QUESTION FOUR (20 Marks)**

Dry saturated steam at a pressure of 95 bar is expanded isothermally to a pressure of 8 bar. Calculate the:

a) Heat supplied

**(10 Marks)**

b) Work done

**(6 Marks)**

c) Show the process on a t-s diagram

**(4 Marks)**

