



(University of Choice)

MASINDE MULIRO UNIVERSITY OF SCIENCE AND TECHNOLOGY (MMUST)

UNIVERSITY EXAMINATIONS 2021/2022 ACADEMIC YEAR (MAIN EXAMINATIONS)

SECOND YEAR SECOND SEMESTER EXAMINATIONS FOR THE DEGREE OF:

BACHELOR OF SCIENCE IN FOOD SCIENCE AND TECHNOLOGY

COURSE CODE: AFS 203

COURSE TITLE: FOOD CHEMISTRY

DATE: 22ND APRIL 2022

TIME: 8.00-10.00AM

Instruction to candidates

ANSWER ALL QUESTIONS IN SECTION A AND ANY TWO IN SECTION B

MMUST observes ZERO tolerance to examination cheating

This paper consists of THREE printed pages. Please Turn Over

SECTION A: ANSWER ALL QUESTIONS

- 1. Using structures, explain why saturated fatty acids have a higher melting points than unsaturated fatty acids (4marks)
- 2. Differentiate between:
- a) Triglycerides and phospholipids
- b) Aromatic and aliphatic amino acids
- c) Cis and Trans fatty acids
- d) Essential and nonessential amino acids (4marks)
- 3. a) Using structures, explain how a tristearate is formed (3marks)
 - b) Give its two physical properties (1marks)
- 4. a) With reference to relevant structures what do you understand by the term asymmetric carbon. (2marks)
 - b) Using the amino acids Serine and threonine below, explain the formation of a peptide bond. (4marks)

- 5. a) What is an amphipathic lipid? (1marks)
 - b) Describe specific functions of steroids (2marks)
 - c) Draw the structure of a fatty acid with 18:2 and give its structural name (3marks)
- 6. a) Using structures, differentiate between starch and cellulose (4marks)
 - b) What terms are used to describe a five and six membered carbon ring monosaccharide respectively (1marks)
 - c) Explain why sucrose is not a reducing agent while maltose is a reducing agent (2marks)
- 7. Differentiate between the following, with examples
 - i) Homopolysaccharide and heteropolysaccharide
 - ii) macro and micro minerals

(2marks)

- b) State any two changes that proteins undergo when they are denatured (2marks)
- 8. a)Briefly describe the four structures of protein

(4marks)

- 9. b) Describe the following in relation to the effect of salt on protein solubility
 - i. Salting-in
 - ii. Salting out

(4marks)

- 10. Illustrate how you would test for protein in a food sample using any of the protein tests (4marks)
- 11. a) Identify the following vitamin (1marks)

b) Explain one physical and one chemical property of the vitamin (2marks)

SECTION B : (20 MARKS) ANSWER ANY TWO QUESTIONS

- 12. a) Discuss two challenges flavourists encounter in flavour chemistry (2marks)
- b) Discuss the flavour perception of food (6marks)
- c) List the 5'ribonucleotide salts (2marks)
- 13. Discuss the physical and chemical properties of carbohydrates and explain their uses in the food industry (10 marks)
- 14. Oxidative rancidity of fats is of great concern to a food processor. Discuss (10marks)
- 15. a) Your mother prepared a fruit salad which turned brown. Explain this scenario. (4marks)
- b) Explain any three ways she would have used to chemically solve the problem? (6marks)

