

MASINDE MULIRO UNIVERSITY OF SCIENCE AND TECHNOLOGY (MMUST)

MAIN CAMPUS

UNIVERSITY EXAMINATIONS 2021/2022 ACADEMIC YEAR

FOURTH YEAR FIRST SEMESTER EXAMINATIONS SUPPLIMENTARY/SPECIAL EXAM

FOR THE DEGREE OF

BACHELOR OF SCIENCE (CHEMISTRY) AND BACHELOR OF EDUCATION (SCIENCE)

COURSE CODE:

SCH 230

COURSE TITLE:

CHEMISTRY OF CARBONYL AND

ORGANOMETALLIC COMPOUNDS

DATE:

28TH JULY 2022

TIME: 8.00 TO 10.00 AM

INSTRUCTIONS TO CANDIDATES

- This paper consists of four questions
- 2. Answer all questions

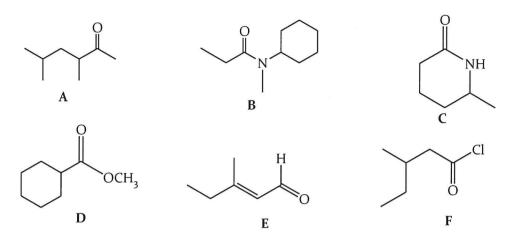
TIME: 2 Hours

MMUST observes ZERO tolerance to examination cheating

This Paper Consists of 5 Printed Pages. Please Turn Over,

Question 1 (16 Marks)

a) Give the systematic names for the compounds whose structures are given below.6 Marks



b) Give structures for the compounds whose IUPAC names are given below.

4 Marks

- i) 2-bromo-4-methoxypentanoic acid
- ii) 5-hydroxyheptan-2-one
- iii) cyclohexyl butanoatecyclohexyl butanoate
- iv) methyl benzenecaboxylate

c) Giving detailed explanation determine which, between each of the two pairs of compounds given below, has a higher boiling point.
i) 6 Marks
ii)

Question 2

Study the reactions given below and the reaction conditions. Giving reasons for your answer determine whether or not the forward reactions are possible.
 5 Marks
 i)

iii)
$$H^+$$
 H_2O H^+ H_2O H^+ $H_3CH_2NH_2$

iv)
$$+ CH_3OH \xrightarrow{HCI} H_3CH_2C \xrightarrow{OCH_3} + OH$$

v)
$$\begin{array}{c} & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ &$$

b) Give products for the following *nucleophilic acyl substitution* reactions and *nucleophilic addition* reactions.

6 Marks

i)
$$OCH_3 + H_2O \stackrel{+}{\longrightarrow} ?$$

iii)

vi)
$$\frac{1. \text{ NaBH}_4}{2. \text{ H}_3\text{O}^+}$$

- c) Refer to question 2, b above to answer the following questions
 - i) What name is given to reaction in question iv) above and how would the forward reaction be enhanced?
 - ii) What difficulties are expected with reaction in question iii) above and how can this be solved? (Note: Give equations to explain this).

 2.5 Marks
 - iii) Give mechanism for the reaction in question v) above

6 Marks

Question 3

a) Arrange the compounds given below in order of reactivity in *nucleophilic addition* reactions starting with the most reactive, then explain the order in each case.
 2 Marks

$$H_{3}C$$
 CH_{3} H H CH_{3}

b) Give products and intermediates for the reactions given below.

i)
$$\begin{array}{c} \stackrel{\text{NH}_2}{\longleftarrow} & + & \stackrel{\text{Excess}}{\longleftarrow} & ? \end{array}$$

iii)
$$O + NH_3 \longrightarrow Pd/C ?$$
Excess

Intermediate

- c) Refer to question 3, b) above to answer the following questions
 - i) On reacting the product obtained in i above with Ag₂O and H₂O compound **X** would be obtained. Give the chemical equation leading to compound **X**? **2** Marks
 - ii) What product do you obtain on exposing compound X to heating? Give the mechanism

3 Marks

iii) Propose a mechanism for reaction v

4 Marks

d) Show how the base catalyst can be used to generate the most stable *enol tautomers* for the compound given below.

2 Marks



Question 4

a) Using an alkyl halide and a thiol as starting materials, how would you prepare the following

thioethers?

CH₃CH₂SCH₂CH₃

B

C

A

Marks

C

D

b) Refer to the reaction scheme below to answer the following questions.

- i) Determine structure of intermediate X 1 Mark
- ii) Give mechanism for the generation of the final acid from intermediate X. 4 Mark
- c) Study the reactions reaction conditions given below and propose the products. 4 Marks

ii)
$$\begin{array}{c}
Bn \\
Bn \\
P \\
Bn
\end{array}
+ CH_3I \xrightarrow{Room} \\
Temperature \\
48 hours$$

iv)
$$CH_2 + Si - CI \xrightarrow{80 \, ^{\circ}C} ?$$