



(University of Choice) MASINDE MULIRO UNIVERSITY OF SCIENCE AND TECHNOLOGY (MMUST)

MAIN CAMPUS

UNIVERSITY EXAMINATIONS 2022/2023 **ACADEMIC YEAR**

FIRST YEAR THIRD TRIMESTER EXAMINATIONS

FOR THE DEGREE OF **BACHELOR OF SCIENCE IN NURSING DISTANCE LEARNING**

COURSE CODE: NCD 131

COURSE TITLE: CLINICAL CHEMISTRY

DATE: Thursday13th April, 2023 TIME: 3:00PM-6:00PM

INSTRUCTIONS TO CANDIDATES

This paper consists of three sections:

- i. Section A – Multiple Choice Questions
- ii. Section B – Short Answer Question
- iii. Section C – Long Answer Question.

Answer all questions

TIME: 2 Hours

MMUST observes ZERO tolerance to examination cheating

This Paper Consists of 4 Printed Pages. Please Turn Over.



Section A: Multiple Choice Questions (20 marks)

Choose the most suitable choice, only one choice is correct

- 1. The enzyme we monitor for heart health found in heart muscle is called: a. Creatinine kinase b. Creatinine c. Troponin d. Creatine kinase 2. A myocardial infarction is the technical term for: a. Inflammation of the myocardium b. Inflammation of the pericardium c. Inflammation of the endocardium d. A heart attack 3. CPK-MB is no longer evident upon testing approximately: a. 3 days later b. 2 days later c. 4 days later d. a week later 4. troponin remains detectable for days following an acute MI a. 14 b. 7 c. 21
- 5. CPK-MB generally appears in just _____hours following an acute MI onset
 - a. 3-6 hours
 - b. 5-8 hours
 - c. 6-8 hours
 - d. 2-4 hours
- 6. An increase in plasma potassium levels is properly called
 - a. hyperpotassemia.
 - b. hyperpotasseplasmia.
 - c. hyperkalemia.
 - d. hypercalcemia.
- 7. Excess potassium ions are eliminated from the body by the
 - a. sweat glands.
 - b. kidneys.
 - c. liver.
 - d. digestive system.
- 8. A hormone that helps to regulate the sodium ion concentration of the blood is
 - a. cortisol.
 - b. parathormone.
 - c. somatotropin.
 - d. aldosterone.
- 9. The most important factor affecting the pH of plasma is the concentration of
 - a. lactic acid.
 - b. organic acids.
 - c. carbon dioxide.
 - d. hydrochloric acid.
- 10. The primary role of the carbonic acid-bicarbonate buffer system is to
 - a. buffer carbonic acid formed by carbon dioxide.
 - b. prevent pH changes caused by organic and fixed acids.

- c. buffer the urine.
- d. increase the amount of carbonic acid during ventilation.
- 11. Prolonged vomiting of the stomach's contents can result in
 - a. respiratory acidosis.
 - b. respiratory alkalosis.
 - c. metabolic acidosis.
 - d. metabolic alkalosis.
- 12. The normal pH range for most body fluids is _____.
 - a. 7 to 8B
 - b. 7.5 to 8
 - c. 7.25 to 7.75
 - d. 7.38 to 7.4
- 13. Abnormal fat and amino acid metabolism may lead to the condition called
 - a. ketoacidosis.
 - b. lactic acidosis.
 - c. metabolic alkalosis.
 - d. respiratory acidosis.
- 14. Normal removal of excess water in urine is known as
 - a. diuresis.
 - b. diuretics.
 - c. osmotic diuresis.
 - d. filtration.
- 15. Creatinine clearance is best used for
 - a. Glomerulonephritis
 - b. Renal failure
 - c. Minor renal impairment
 - d. Cystitis
- 16. To measure glomerular filtration you need a substance that is
 - a. Has a limited filtration
 - b. Exogenous
 - c. Endogenous
 - d. Reabsorbed by renal tubules
- 17. Normal range of serum urea in adult is
 - a. 2.5-7.7 mmol/L
 - b. 3.6-6.6 mmol/L
 - c. 3.6-8.8 mmol/L
 - d. 2.5-6.6 mmol/L
- 18. Which of the following marker is used for the differential diagnosis of obstructive jaundice?
 - a. Lactate dehydrogenase
 - b. Creatine Kinase
 - c. Carbonic anhydrase
 - d. 5'- Nucleotidase
- 19. All of the following statement is true concerning urobilinogen, Except:
 - a. Produced by oxidative action of intestinal bacteria.
 - b. Undergoes significant enterohepatic circulation.
 - c. Urinary levels increased in biliary obstruction.
 - d. Fecal levels decreased in biliary obstruction.
- 20. Galactosemia is caused by a deficiency of the enzyme responsible for galactose metabolism. Name the enzyme. Please select one correct answer from the following options:
 - a. Galactokinase
 - b. Galactose 1- phosphate Uridyl transferase

- c. UDP-4- Epimerase
- d. Galactose dehydrogenase

Section B Short Answer Question (40 marks)

- 1. Explain the renal acid base control mechanisms (8marks)
- 2. Explain the key indication for renal tests (8marks)
- 3. Explain the potential uses of tumor makers
- 4. Describe the body water compartment and associated electrolytes (8marks).
- 5. Outline the common cardiac biomarker (8marks).

Section C Long Answer Questions (40 marks)

- 1. 77 year-old male patient came to the clinic with sudden lower back pain, a serum creatinine analysis was done and the serum creatinine level was 0.55mg/dL, calculate the glomerular filtration rate knowing that the patient's body weight is 63kg. (K= 1.23, to convert mg/dL to µmol/L we need to multiply by 88.4) (20marks)
- 2. Discuss nursing management of patient with potassium imbalances (20marks)