



(University of Choice)

MASINDE MULIRO UNIVERSITY OF SCIENCE AND TECHNOLOGY (MMUST)

MAIN CAMPUS

UNIVERSITY EXAMINATIONS

2021/2022 ACADEMIC YEAR

SECOND SEMESTER MAIN EXAMINATION

FOR THE DEGREE OF BACHELOR OF SCIENCE IN ELECTRICAL AND COMMUNICATION (2ND YEAR)

COURSE CODE: MIE 274:

COURSE TITLE: THERMODYNAMICS I

DATE: 29-04-2022

TIME: 15:00-17:00

INSTRUCTIONS:

- 1. This paper contains FOUR questions
- 2. QUESTION ONE IS COMPULSORY
- 3. Attempt any TWO questions from the remaining.
- 4. Question ONE carries 30 marks and the REST 20 marks each.
- 5. Examination duration is 2 (TWO) HOURS

MMUST observes ZERO tolerance to examination cheating

This Paper Consists of 3 Printed Pages. Please Turn Over.

QUESTION ONE (30Marks)

- a) State and discuss five gas laws you are aware of in reference to thermodynamics (10 Marks)
- b) 10Kg of fluid per minute goes through a reversible steady flow process. The properties of the fluid at the inlet are: Pressure, P₁= 1.5 bar; density, ρ₁= 26Kg/m³; velocity, C₁ = 110m/s and internal energy, u₁= 910KJ/Kg and at exit are: Pressure, P₂= 5.5 bars; density, ρ₂= 5.5Kg/m³; velocity, C₂ = 190m/s and internal energy, u₂= 710KJ/Kg. During the passage, the fluid rejects 55KJ/s and rises through 55 metres. Take gravitational force, g = 9.80 m/s². Determine:
 - (i) The change in enthalpy(4 Marks)
 - (ii) Work done during the process. (6 Marks)
- c) A 2200-kg automobile traveling at 25 m/s hits the rear of a stationary, 1000-kg automobile. After the collision the large automobile slows to 13.89 m/s and the smaller vehicle has a speed of 24.44 m/s. What has been the increase in internal energy, taking both vehicles as the system? (10 Marks)

QUESTION TWO (20Marks)

- a) An insulated vessel of 0.5 m³ capacity is divided by a rigid conducting diaphragm into two chambers A and B, each having a capacity of 0.25 m³. Chamber A contains air at 1.4 bar pressure and 290 K temperature and the corresponding parameters for air in chamber B are 4.2 bar and 440 K. Calculate:
- (i) Final equilibrium temperature, (8 Marks)
- (ii) Final pressure on each side of the diaphragm (2 Marks).

For air take $c_v = 0.715 \text{ kJ/kg K}$ and R = 0.287 kJ/kg K.

- b) 1 Kg of steam initially dry saturated at 1.1 Mpa expands in a cylinder following the law $PV^{1.13} = C$. The pressure at the end of expansion is 0.1 Mpa. Determine:
 - i. The final volume (2 Marks).
 - ii. Final dryness fraction (2 Marks).
- iii. Work done (2 Marks).
- iv. The change in internal energy (3 Marks).
- v. The heat transferred (1 Marks).

QUESTION THREE (20Marks)

- a) A heat engine receives heat at the rate of 1,500KJ/s and gives an output of 8.2KW. Determine:
- (i) The thermal efficiency (2Marks)
- (ii) The rate of heat rejection (2Marks)
- b) What amount of heat would be required to produce 4.4Kg of steam at a pressure of 6 bar and temperature of 250°C from water at 30°C? Take specific heat for water as 4.18 kJ/kg K and superheated steam as 2.2KJ/KgK. (10 Marks)
- c) Define the following terms in reference to Thermodynamics. (6 Marks)
- (i) Mass Density
- (ii) Specific Weight
- (iii) Specific Volume

QUESTION FOUR (20Marks)

- a) Steam at a 6.87 bar, 205°C, enters in an insulated nozzle with a velocity of 50 m/s. It leaves at a pressure of 1.37 bar and a velocity of 500 m/s. Determine the final enthalpy of steam (8 Marks).
- b) State the rules governing non-flow systems in thermodynamics (3 Marks).
- c) Air at 1.02 bar, 22°C, initially occupying a cylinder volume of 0.015 m³, is compressed reversibly and adiabatically by a piston to a pressure of 6.8 bar. Calculate:
 - i. The final temperature (3 Marks)
 - ii. The final volume (3 Marks)
- iii. The work done on the mass of air in the cylinder (3 Marks)

