

(University of Choice)

MASINDEMULIROUNIVERSITY OF SCIENCE AND TECHNOLOGY (MMUST)

UNIVERSITY MAIN EXAMINATIONS 2020/2021 ACADEMIC YEAR

FIRST YEAR, THIRD TRIMESTER EXAMINATION FOR THE DEGREE OF BACHEOR OF SCIENCE IN CLINICAL MEDICINE, SURGERY & COMMUNITY HEALTH

COURSE CODE:

HCM130

COURSE TITLE:

MEDICAL BIOCHEMISTRY III

DATE: WEDNESDAY 4TH AUGUST 2021TIME:8.00-11.00 AM

INSTRUCTIONS TO CANDIDATES

Answer All Questions

Section A: Multiple Choice Questions (MCQs)

Section B: Short Answer Questions (SAQs)

Section C: Long Answer Question (LAQs)

20 Marks.

40 Marks.

40Marks

Time: 3 hours

MMUST observes ZERO tolerance to examination cheating

This Paper Consists of 5 Printed Pages, Please Turn Over.

Section A: Multiple Choice Questions (MCQs) (20MARKS)

- 1. Quantitatively, the most significant buffer system in plasma comprises:
 - A. Disodium hydrogen phosphate/sodium dihydrogen phosphate
 - B. Carbonic acid-bicarbonate
 - C. Plasma proteins
 - D. Lactic acid-lactate buffer
- 2. Which of the following physiological/pathological conditions is most likely to result in an alkalosis, provided the body could not fully compensate:
 - A. Production of lactic acid by muscles during exercise
 - B. Production of ketone bodies by a patient with diabetes mellitus
 - C. Repeated vomiting of stomach contents, including HCL
 - D. Diarrhoea with loss of bicarbonate anions secreted into the intestine
- 3. A patient with an enteropathy (intestinal disease) produced large amounts of ammonia (NH₃) from bacterial overgrowthin the intestine. The ammonia was absorbed through the intestine into the portal vein and entered the circulation. Which of the following is a likely consequence of his ammonia absorption:
 - A. A decrease of blood pH
 - B. Conversion of ammonia to ammonium ion in the blood
 - C. Increased expiration of CO₂
 - D. A decreased concentration of bicarbonate in the blood
- 4. Chloride shift involves:
 - A. H⁺ ions leaving the RBCs in exchange of Cl
 - B. Cl- leaving the RBCs in exchange of bicarbonate
 - C.Bicarbonate ion returns to plasma and exchanged with chloride which shifts into the cell
 - D. Carbonic acid to the plasma
- 5. Anion gap is the difference in the plasma concentrations of:
 - A. (Chloride) (Bicarbonate)
 - B. (Sodium) -(Chloride)
 - C. (Sodium + Potassium) (Chloride + Bicarbonate)
 - D. (Sum of Cations) (Sum of Anions)
- 6. The isozyme CK-MB is specifically increased in the blood of patients who have:
 - A. Skeletal muscle disease
 - B. Infective hepatitits
 - C. Myxoedema
 - D. Recent myocardial infarction
- 7. Which of the following enzyme is typically elevated in alcoholism:
 - A. Serum ALP
 - B. Serum GOT

- C. Serum γ-GT
- D. Serum Acid Phosphatase

8. The best test for acute pancreatitis in the presence of mumps is:

- A. A serological test for mumps
- B. Serum amylase
- C. Urinary amylase
- D. Serum lipase

9. Which of the following is known as bone forming enzyme:

- A. Acid phosphatase
- B. Alkaline phosphatase
- C. Lactate dehydrogenase
- D. γ-glutamyl transpeptidase

10. The blood sugar raising action of the hormone of suprarenal cortex is due to:

- A. Glyconeogenesis
- B. Glycogenolysis
- C. Glucagon-like activity
- D. Inhibition of glomerular filtration of glucose

11. Mineralocorticoids increase the tubular re-absorption of :

- A. Sodium and calcuim
- B. Sodium and potassium
- C. Sodium and chloride
- D. Potassium and chloride

12. High levels of T₃ and T₄ and low TSH in serum indicates:

- A. Hyperthyroidism of pituitary origin
- B. Hypothyroidism of pituitary origin
- C. Hyperthyroidism of thyroid origin
- D. Hypothyroidism of thyroid origin

13. A characteristic of pheochromocytoma is elevated urinary excretion of:

- A. Dopamine
- B. Tyrosine
- C. Phenylalanine
- D. Vinylmandelic acid

14. A high concentration of PTH in blood causes:

- A. Decrease in plasma calcium and increase in plasma inorganic phosphorous
- B. Increase in plasma calcium and decrease in plasma inorganic phosphorous
- C. Increase in plasma calcium and inorganic phosphorous
- D. Decrease in plasma calcium and inorganic phosphorous

15. Serum inorganic phosphorous decreases in all the following conditions except:

- A. Hyperparathyroidism
- B. Intestinal malabsorption
- C. Osteomalacia
- D. Chronic renal failure

16. Hypocalcaemia can occur in all the following except:

- A. Rickets
- B. Osteomalacia
- C. Hyperparathyroidism
- D. Intestinal malabsorption

17. Iron is stored in the form of:

- A. Ferritin and transferrin
- B. Transferrin and haemosiderin
- C. Haemoglobin and myoglobin
- D. Ferritin and haemosiderin

18. In iron deficiency anemia:

- A. The plasma bound iron is high
- B. The plasma bound iron is low
- C. Total iron binding capacity is low
- D. Both the plasma bound iron and total iron binding capacity are low

19. Diets having a high ratio of polyunsaturated: saturated fatty acids can cause:

- A. Increase in serum triglycerides
- B. Decrease in serum HDL
- C. Decrease in serum cholesterol
- D. Skin lesions

20. Action of insulin on lipid metabolism is:

- A. It increases lipolysis and increases triglyceride synthesis
- B. It decreases lipolysis and increases triglyceride synthesis
- C. It decreases lipolysis and decreases triglyceride synthesis
- D. It increases synthesis of triglyceride and increased ketogenesis

Section B: Short Answer Questions (SAQs) (40 MARKS)

- 1. Briefly explain any five (5) biochemical markers reflecting cardiac tissue damage and myocardial function(10 marks)
- 2. Using appropriate examples, briefly explain the classification of renal function tests (RFT's)(10 marks)
- 3. Using suitable examples briefly explain at least five (5) main clinical parameters utilized in the classification of tumor markers for cancer diagnosis (10 marks)
 - 4. Briefly explain the clinical (biochemical) test parameters and findings (interpretation) of Iron status in disease (10 marks)

Section C: Long Answer Questions (LAQs) (40 MARKS)

- 1. a) Briefly describe five (5) major factors which regulate renal bicarbonate (HCO3⁻) reabsorption and acid (H⁺) excretion (10 marks)
 - b) Briefly explainany five (5) blood chemistry components of an arterial blood gas analysis (BGA) profile (10 marks)
- a)List atleast five (5) different biochemical analytes that can be measured in cerebrospinal fluid (CSF), while explaining their clinical application/significance(10 marks)
 - b) Providing relevant examples, briefly perform the classification of liver function tests (LFTs) based on the specific functions of the liver involved(10 marks)

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SECTION C

- 1) Germander GIFR
- ii) Ammonium
- 111) PCO2.
- (v) Extracellular fluid volume
- V) Absence of potassium & chloride