



### MASINDE MULIRO UNIVERSITY OF SCIENCE AND TECHNOLOGY

#### UNIVERSITY EXAMINATIONS 2021/2022 ACADEMIC YEAR

## SECOND YEAR SPECIAL/SUPPLEMENTATRY EXAMINATION

#### FOR THE DEGREE OF

#### **BACHELOR OF INDUSTRIAL CHEMISTRY**

COURSE CODE:

SCH 261

COURSE TITLE: CHEMISTRY OF FUNCTIONAL GROUPS

THURSDAY,  $28^{TH}$  JULY 2022 DATE:

TIME: 11.00 - 1.00PM

#### **INSTRUCTIONS**

Answer ALL Questions as directed.

MMUST observes ZERO tolerance to examination cheating

This paper consists of 3 printed pages. Please turn over.

## **QUESTION 1**

(25 marks)

[3]

- (a) Explain the meaning of the term "Functional Group (FG)" as used in Chemistry.
- (b) You are given an important organic molecule with the chemical structure shown below. Study the structure carefully and answer the following questions.

- (i) Identify all the functional groups in the molecule (Each type of FG to be reflected once only) [10]
- (ii) Of the FGs in (i) above, based on structure only, which one is of the highest priority? [2]
- (iii) How many  $sp^2$  hybridized and tertiary carbons are found in the molecule? [2]
- (iv) Name five (5) types of reactions this molecule may undergo? [5]
- (v) Speculate and cite the possible function(s) of the above molecule based purely on the chemical structure given. [3]

## **QUESTION 2**

(14 marks)

(a) With reasons, classify the following reactions as "Substitution" or "Elimination". [8]

a) Br 
$$\frac{H_2O}{i\text{-PrOH}}$$
 OH

d) 
$$C_8H_8CH_2Br$$
  $C_8H_8CH_2OCH_3$ 

# (b) Give IUPAC names of the following molecules.

## (c) Give the products of the following reactions.

QUESTION 3 (17marks)

- (a) Oxidation of ketones to carboxylic acid esters using a peroxy-acid as the oxidizing agent is given a special name. State the name of this reaction. [2]
- (b) Using your own example a ketone and a peroxy-acid, outline the reaction mechanism of the reaction in (a) above. [15]

QUESTION 4 (14 marks)

(a) What is the IUPAC name of the following carboxylic acid? [1]

- (b) Indicate on the structure above  $\alpha$ ,  $\beta$ ,  $\gamma$ ,  $\delta$  and  $\epsilon$  carbon atoms. [5]
- (c) Explain why melting points and boiling points of carboxylic acids generally increase with increase in number of carbons / molecular mass. [2]
- (d) List six (6) functions of carboxylic acids and their derivatives [6]