

(University of Choice)

# MASINDE MULIRO UNIVERSITY OF SCIENCE AND TECHNOLOGY (MMUST)

**MAIN CAMPUS** 

**UNIVERSITY EXAMINATIONS** 

**2022/2023 ACADEMIC YEAR** 

FIRST YEAR SECOND SEMESTER EXAMINATIONS

FOR THE DEGREE

**OF** 

MASTER OF SCIENCE (CHEMISTRY)

**COURSE CODE:** 

**SCH 843E** 

COURSE TITLE:

ADVANCED ELECTROCHEMISTRY

DATE: 13-4-2023

TIME: 9-12 P.M.

## INSTRUCTIONS TO CANDIDATES

Answer all the Questions

TIME: 3 Hours

MMUST observes ZERO tolerance to examination cheating

This Paper Consists of 4 Printed Pages. Please Turn Over.

### **QUESTION ONE (15 MARKS)**

a. What is the significance of the gradient of Heyrovsky -Ilkovic graph?

(5 marks)

b. Consider the reduction at a DME of Fe  $^{2+}$  to form Fe. What is the accurate value of  $E_{1/2}$  from the data shown below? Take  $I_d = 3.24 \,\mu A$  and assume a temperature of 20°C. (10 marks)

E/V vs SCE	-0.395	-0.406	-0.415	-0.422	-0.431	-0.445
Ι/μΑ	0.48	0.97	1.46	1.94	2.43	2.92

#### **QUESTION TWO (15 MARKS)**

- a. The one-electron reduction of the heptyl viologen dication ( $HV^{2+}$  occurs at a platinum electrode with an area of 1.50 cm<sup>2</sup>. Calculate the value of ket, given that the intercept on a Tafel plot is -2.2. Comment on the answer obtained (5 marks)
- b. One way to prevent iron pipe from rusting is to connect it with a piece of wire to a magnesium rod. What is the electrochemical principle of this action (3 marks)
- c. A solution containing potassium ions is known to also contain NaCl at a concentration of  $0.001 \text{ mol dm}^{-3}$ . When  $0.02 \text{ mol dm}^{-3}$  KNO<sub>3</sub> is dissolved in the same chloride solution, the emf is 1.403 V. What is the concentration of K<sup>+</sup> in a new sample if the emf is 1.390 V) (7 marks)

#### **QUESTION THREE (15 MARKS)**

Is non faradaic current a problem in polarography? Discuss this statement with possible solutions and explanations. Use well illustrated diagrams. (15 marks)

# **QUESTION FOUR (15 MARKS)**

- a. Compare and Contrast the Tafel and Butler Volmer approach to electrode kinetics, outlining clearly the advantages and disadvantages of each approach (10 marks)
- b. Apply the Butler-Volmer equation to determine the net current at an overpotential  $\eta$  of 0.120 V (Take  $\alpha$ =0.5T, T=298K, n=1 and I $_{o}$ =0.034 A). (5 marks)